AIPMT SAMPLE PAPERS WITH SOLUTIONS

PHYSICS

Choose the correct answer :

- 1. Which one of the following quantities has not been expressed in proper units?
 - (1) $\frac{\text{Stress}}{\text{Strain}} = \frac{\text{N}}{\text{m}^2}$
 - (2) Surface tension = $\frac{N}{m}$
 - (3) Energy = kgm/s
 - (4) Pressure = $\frac{N}{m^2}$
- 2. If C and L denote the capacity and inductance, then the dimension of LC are
 - (1) $M^0L^0T^2$ (2) ML^2T^{-2}
 - (3) $M^2L^2T^{-2}$ (4) MLT^{-1}

- 3. For the resultant of two vectors to be maximum, what must be the angle between them?
 - (1) 0° (2) 60°
 - (3) 90° (4) 180°
- 4. Two blocks of masses 2 kg and 1 kg are in contact with each other on a frictionless table when a horizontal force of 3 N is applied to the block of mass 2 kg, the value of the force of contact between the two blocks
 - (1) 4 N
 - (2) 3 N
 - (3) 2 N
 - (4) 1 N

5. Torque acting on rod about an axis, as shown, is



- (1) 5 N m (2) $5\sqrt{3} \text{ N m}$
- (3) 10 N m (4) $10\sqrt{3}$ N m
- A body of mass 2 kg moving with a velocity of 3 m/s collides head on with a body of mass 1 kg moving with a velocity of 4 m/s in opposite direction. After collision the two bodies stick together and move with a common velocity

(1)
$$\frac{1}{4}$$
 m/s
(2) $\frac{1}{3}$ m/s
(3) $\frac{2}{3}$ m/s
(4) $\frac{3}{4}$ m/s

- 7. Two skaters *A* and *B* of masses 50 kg and 70 kg respectively stand facing each other 6 m apart. They pull the rope stretched between them. How far has each moved when they meet?
 - (1) Both have moved 3 m
 - (2) Both have moved 2.5 m
 - (3) A moves 3.5 m and B 2.5 m
 - (4) A moves 2 m and B 4 m
- 8. A particle is moving in a straight line has velocity, displacement equation as $v = \sqrt{1+s}$, where v is in m/s and s is in m. The initial velocity of the particle

(1) 2 m/s	(2) 1 m/s
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- (3) 3 m/s (4) 4 m/s
- 9. A weight is suspended from the roof of a lift by a spring balance. When the lift is stationary the spring balance reads *W*. If the lift suddenly falls freely under gravity the reading of the spring balance will be
 - (1) *Wg* (2) 2*Wg*
 - (3) $\frac{W}{2}g$ (4) Zero

10. The kinetic energy *E* of a particle of mass *m* moving in a circle of radius *r* varies with distance traced *S* as $E = 4KS^2$. The tangential acceleration of the particle, is

(1)
$$\frac{4K}{m}$$
 (2) $\frac{8KS}{m}$
(3) $\frac{2K}{m}$ (4) $\frac{8KS^2}{mr}$

- 11. A fan of moment of inertia 1.6 kg m² is to be run upto a working speed of $\frac{1}{2}$ rps. Correct value of the angular momentum of the fan, is
 - (1) $1.6 \pi \text{ kg m}^2/\text{s}$ (2) $2.6 \pi \text{ kg m}^2/\text{s}$
 - (3) $6 \text{ kg m}^2/\text{s}$ (4) $3 \text{ kg m}^2/\text{s}$
- 12. A simple pendulum has an amplitude 2A and time -

period $\frac{T}{4}$. The max velocity will be



13. Which of the following graphs represents the motion of a planet moving around the sun?



- 14. Two pieces of metal when immersed in a liquid have equal upthrust on them, then
 - (1) Both the pieces must have equal weights
 - (2) Both the pieces must have equal densities
 - (3) Both the pieces must have equal volume
 - (4) Both are floating to the same depth

15. A copper wire of length 1 m and cross-section $\frac{1}{16}$ m²

stretched by a load of 2 kg. Young's modules for copper is 2 × 10^{11} N/m² and g = 10 m/s². The extension of the wire

- (1) 1.6×10^{-10} m (2) 16×10^{-10} m
- (3) 8×10^{-8} m (4) 16×10^{-11} m
- 16. The speed of sound in air is
 - (1) \propto pressure of air
 - (2) \propto square of pressure
 - (3) $\propto \sqrt{\text{pressure}}$
 - (4) Independent of pressure if temperature is constant
- 17. An open organ pipe has fundamental frequency of 400 Hz. The frequency of first overtone of this pipe is same as that of 1st overtone of a closed organ pipe. The length of the closed organ pipe is nearly
 - (1) 31 cm (2) 39 cm
 - (3) 28 cm (4) 34 cm
- Two cylinders of same diameter one of iron and other of silver, are placed in closed contact as shown in figure. If K_{Iron} = 12 K_{Ag} what will the temperature of the interface nearly?



(1) 92°C

- (2) 94°C
- (3) 95°C
- (4) 96°C

19. If 2 moles of a monatomic gas $\gamma = \frac{5}{3}$ is mixed with

1 mole of diatomic gas $\gamma = \frac{7}{5}$, the value of γ for the mixture is

- (1) $\frac{17}{11}$ (2) $\frac{11}{17}$
- (3) $\frac{16}{11}$ (4) $\frac{11}{16}$

- 20. In a particular experiment, a gas undergoes adiabatic expansion, satisfying the equation VT^3 = constant. The ratio of specific heats is equal to
 - (1) 4 (2) 3

(3)
$$\frac{5}{3}$$
 (4) $\frac{4}{3}$

21. A concave mirror of focal length *f* (in air) is immersed in water $\left(\mu = \frac{4}{3}\right)$. The focal length of mirror in water will be

(1)
$$f$$
 (2) $\frac{4}{3}f$

(3)
$$\frac{3}{4}f$$
 (4) $\frac{7}{3}f$

- 22. A convex lens and a concave lens, each having same focal length of 30 cm are put in contact to form a combination of lenses. The power in diopters of the combination is
 - (1) Zero (2) 25
 - (3) 50 (4) Infinite
- The refractive indices of violet and red light are
 1.54 and 1.52 respectively if the angle of prism is
 10°. The angular dispersion is
 - (1) 0.02° (2) 0.2°
 - (3) 3.06° (4) 30.6°
- 24. The least distance of distinct vision for a long-sighted person is *x* m. He wants to read a newspaper placed

at a distance of $\left(\frac{x}{2}\right)$ m. The power of the spectacle

lens used by him is x diopter. What is the value of x?

- (1) 0.5 (2) 1
- (3) 1.5 (4) 2
- 25. The fringe width in Young's double slit experiment increases when
 - (1) Wavelength increases
 - (2) Distance between source and screen decreases
 - (3) Distance between slit increases
 - (4) The width of the slit increases

- 26. If a spherical hollow conductor encloses a charge *q*, then the total flux coming out of the surface will be
 - (1) $\frac{1}{\varepsilon_0}$ × the charge enclosed by the surface
 - (2) $\varepsilon_0 \times$ charge enclosed by surface

(3)
$$\frac{1}{4\pi\varepsilon_0}$$
 × charge enclosed by surface

- (4) Zero
- 27. Two point charges q and -q are at a positions (0, 0, d) and (0, 0, -d) respectively. What is the electric field at (a, 0, 0)?

(1)
$$\frac{2qd}{4\pi\varepsilon_0 (a^2 + d^2)^{3/2}} \hat{k}$$

(2)
$$\frac{qd}{4\pi\varepsilon_0 (a^2 + d^2)^{3/2}} \hat{k}$$

(3)
$$\frac{-2qd}{4\pi\varepsilon_0 (a^2 + d^2)^{3/2}} \hat{k}$$

(4)
$$\frac{-qd}{4\pi\varepsilon_0 (a^2 + d^2)^{3/2}} \hat{k}$$

28. A circuit has a section AB and shown in the figure. If the potential difference between points A and B is V volt, then the potential difference across C_1 is



(4)
$$\frac{(V-E)C_2}{C_1+C_2}$$

29. In the given figure the current through 4 ohm resistor is



30. In the circuit shown below, if the resistance $5\,\Omega$ develops heat of 45 J/s, heat developed in $2\,\Omega$ must be about



31. An insulating rod of *L* carries a charge *Q* distributed uniformly on it. The rod is pivoted at on end and is rotated at a frequency *f* and about a fixed perpendicular axis. The magnetic dipole moment of the system is

(1)
$$\frac{\pi Q}{L^2}$$
 (2) $\frac{QL^2 \pi f}{3}$
(3) $QL\pi f$ (4) $\frac{QL^2}{3}$

- 32. Relative permittivity and permeability of a material are ε_r and μ_r , respectively which of the following values of these quantities are allowed for a diamagnetic material?
 - (1) $\varepsilon_r = 0.5, \ \mu_r = 1.5$ (2) $\varepsilon_r = 1.5, \ \mu_r = 0.5$
 - (3) $\varepsilon_r = 0.5, \ \mu_r = 0.5$ (4) $\varepsilon_r = 1.5, \ \mu_r = 1.5$
- 33. A current of 2 A flows through an inductor of inductance 0.1 H. What is the maximum magnetic energy stored in the inductor ?
 - (1) 0.2 J (2) 0.1 J
 - (3) 0.4 J (4) 0.8 J

34. A sinusoidal voltage $V_0 \sin \omega t$ is applied across a series combination of resistance *R* and inductance *L*. The amplitude of the current in the circuit is

(1)
$$\frac{V_0}{\sqrt{R^2 + \omega^2 L^2}}$$
 (2) $\frac{V_0}{\sqrt{R^2 - \omega^2 L^2}}$
(3) $\frac{V_0}{R + \omega L}$ (4) $\frac{V_0}{R}$

35. A plane electromagnetic wave,

 $E = 200 \sin(2 \times 10^8 t + 1x)$ V/m propagates in a medium of refractive index

- (1) 1.5 (2) 2.0
- (3) 10 (4) 2.4
- 36. An α particle is accelerated through a potential difference of 200 V. The increase in its kinetic energy, in electron volt is
 - (1) 100 (2) 200
 - (3) 400 (4) 800
- 37. The threshold frequency for a certain metal v_0 . When light of frequency $v = 2v_0$ is incident on it, the maximum velocity of photoelectrons is 4×10^6 m/s. If the frequency of incident radiation is increased to $5v_0$, then the maximum velocity of the photoelectron (in m/s) will be

(1)
$$\frac{4}{5} \times 10^{6}$$
 (2) 2×10^{6}
(3) 8×10^{6} (4) 2×10^{7}

- The angular speed of the electron in the *n*th orbit of Bohr's hydrogen atom is
 - (1) Directly proportional to n
 - (2) Inversely proportional to \sqrt{n}
 - (3) Inversely proportional to n^2
 - (4) Inversely proportional to n^3
- 39. The half life of a radioactive element which has only

 $\frac{1}{32}$ of its original mass left after a lapse of 60 days, is

- (1) 12 days (2) 32 days
- (3) 60 days (4) 64 days

- 40. Fusion reaction takes place at about
 - (1) 3 × 10² K
 - (2) $3 \times 10^3 \text{ K}$
 - (3) 3 × 10⁴ K
 - (4) 3 × 10⁷ K
- 41. In a P-N junction diode, holes diffuse from *P* region to *N* region because
 - (1) The force electrons in the N region attract them
 - (2) They are swept across the junction by potential difference
 - (3) There is greater concentration of holes in P region as compared to N region
 - (4) Diffuse only when there is reverse biasing
- 42. The difference in the variation of resistance with temperature in a metal and a semiconductor arises essentially due to the difference in
 - (1) Type of bonding
 - (2) Crystal structure
 - (3) Scattering mechanism with temperature
 - (4) Number of charge carries with temperature
- 43. Current flowing in each of the circuit A and B respectively



- 44. The process of superposition of weak audio signal to a high frequency carrier wave is called
 - (1) Transmission (2) Communication
 - (3) Modulation (4) Demodulation
- 45. A TV tower has a height of 50 m. The maximum distance upto which TV transmission can be received is approximately equal to (R = 6400 km)
 - (1) 5 km (2) 25 km
 - (3) 100 km (4) 250 km
- (5)

CHEMISTRY

- 46. Which of the following set contains only intensive variables?
 - (1) Volume, energy, enthalpy
 - (2) Area, enthalpy, work
 - (3) Viscosity, surface tension, density
 - (4) Viscosity, density, volume
- For a reaction, N₂(g) + 3H₂(g)→2NH₃(g), the rate of formation of NH₃ is 34 g hour⁻¹. The rate at which H₂ is consumed, under similar conditions is
 - (1) 12 g/hr (2) $\frac{4}{3}$ g/hr (3) 6 g/hr (4) $\frac{1}{2}$ g/hr
- 48. The number of moles of $KMnO_4$ which is required to react with one mole of sulphite ion (SO_3^{2-}) in acidic medium is
 - (1) $\frac{4}{5}$ (2) $\frac{3}{5}$ (3) 1 (4) $\frac{2}{5}$
- Fe³⁺, Zn²⁺ and Cu²⁺ ions are present in a slightly acidic medium. The reagent which when added in excess would separate Fe³⁺ in one step is

(1)	H ₂ S	(2)	HCI
(3)	NaOH	(4)	NH_3

50. How many pairs of enantiomers are possible for the molecular formula C₅H₁₁Br?

(1)	2	(2)	3
(3)	4	(4)	5

- 51. The type of isomerism present in $[Cr(NH_3)_5 NO_2]^{2+}$ is
 - (1) Ionisation isomerism
 - (2) Optical isomerism
 - (3) Linkage isomerism
 - (4) Geometrical isomerism
- 52. Which of the following will not react with acetylene?
 - (1) NaNH₂
 - (2) Tollen's reagent
 - (3) HCI
 - (4) Both (2) & (3)

- 53. Equal weights of two gases separately expands in an isothermal and reversible manner and the work done in both the cases is same. The gases are
 - (1) F_2, O_2 (2) N_2, C_2H_4
 - (3) CH_4 , NO₂ (4) C_2H_6 , N₂O
- 54. The volume of 0.1 M of KOH required to neutralise completely 20 ml of 0.1 M H₃PO₃, is
 - (1) 50 ml (2) 40 ml
 - (3) 30 ml (4) 20 ml
- The coordination number and oxidation number of Cr in K₃[Cr(C₂O₄)₃] are respectively
 - (1) 6 and +3 (2) 3 and +6
 - (3) 4 and +2 (4) 3 and +3
- 56. 2-Methylpentan-2-ol is prepared from acetone and 'x'. '*x*' is
 - (1) (CH₃)₃CMgl/H₂O
 - (2) C₂H₅MgCl/H₂O
 - (3) CH₃CH₂CH₂MgCl/H₂O
 - (4) CH₃MgCl/H₂O
- 57. The conjugate acid of formic acid in the given reaction is

 $HCI + HCOOH \rightleftharpoons HCOOH_2 + CI^-$

- (1) CI⁻ (2) HCOO⁻
- (3) $HCOOH_2$ (4) HCOOH
- 58. Consider the following equimolal solutions
 - a. Sugar b. Sodium chloride
 - c. Sodium sulphate d. Sodium phosphate

The correct order of increasing freezing point of the given solutions is

- (1) a < b < c < d (2) b < c < a < d
- (3) d < c < b < a (4) c < d < a < b
- 59. 0.5 ampere current is passed through copper and silver voltameters for 40 seconds. The metal that gets deposited more in gram is
 - (1) Cu
 - (2) Ag
 - (3) Both in equal quantities
 - (4) Cannot be predicted

- 60. In a cubic crystal anions are arranged in fcc and cations occupy half the tetrahedral voids and all the octahedral voids. The ratio of cations and anions in the crystal is
 - (1) 3:2
 (2) 1:2

 (3) 2:3
 (4) 2:1
- 61. Alcohols reacts with Na according to the equation

 $2ROH + 2Na \rightarrow 2RONa + H_2$

The order of reactivity of different alcohols towards Na is

- (1) $3^{\circ} > 2^{\circ} > 1^{\circ}$ (2) $1^{\circ} > 2^{\circ} > 3^{\circ}$
- (3) $2^{\circ} > 3^{\circ} > 1^{\circ}$ (4) $3^{\circ} > 1^{\circ} > 2^{\circ}$
- 62. The reagent which reacts differently with HCHO, CH_3CHO and CH_3COCH_3 is
 - (1) NH₂OH (2) NH₃
 - (3) HCN (4) Both (2) & (3)
- 63. The correct statement/s is/are
 - a. Glucose contains 3 chiral centres in its hemiacetal form
 - b. Glucose is an aldohexose
 - c. Naturally occurring glucose is dextrorotatory
 - d. Glucose contains one 2° alcoholic group and three 1° alcoholic group
 - (1) Both a & b (2) Both b & c
 - (3) a, b & c (4) All are correct
- 64. Which of the following can be hydrolysed most readily?

(1)
$$CH_{3} - C - CI$$

(2) $CH_{3} - C - O - C - CH_{3}$
(3) $CH_{3} - C - OC_{2}H_{5}$

(4) $CH_3 - C - NH_2$

- 65. Which statement about pyridine is correct?
 - (1) It is more basic than ethylamine
 - (2) It cannot act as a good nucleophile
 - (3) It is less basic than pyrrole
 - (4) All are correct

66. The equilibrium constant of the reaction

 $HONO + CN^{-} \rightleftharpoons HCN + ONO^{-}$ is 1.1×10^{-4}

Considering the reaction we can conclude that

- (1) CN⁻ ion is a weaker base than ONO⁻
- (2) HCN is a weaker acid than HONO
- (3) NO⁻ is the conjugated base of HONO
- (4) The conjugate acid of CN⁻ is HCN
- 67. If the activation energy of a reaction is zero, then the rate constant (k) of the reaction
 - (1) Decreases with increase in temperature
 - (2) Decreases with decrease in temperature
 - (3) Increases with increase in temperature
 - (4) Is nearly independent of temperature
- 68. Which one of the following compounds will not be oxidised by acidified potassium dichromate?
 - (1) (CH₃)₂CHOH (2) (CH₃)₃COH
 - (3) CH₃OH (4) CH₃CH₂OH
- 69. Polystyrene, dacron and orlon are classified respectively as
 - (1) Chain growth, step growth, step growth polymer
 - (2) Chain growth, step growth, chain growth polymer
 - (3) Step growth, step growth, chain growth polymer
 - (4) Step growth, chain growth, step growth polymer
- 70. When 0.1 mole of urea is dissolved in 10 moles of water, the vapour pressure of water will be
 - (1) Increased by 10% (2) Decreased by 10%
 - (3) Decreased by 1% (4) Increased by 1%
- 71. Acetoxime on reduction followed by acetylation gives
 - (1) Isopropylamine
 - (2) Ethylamine
 - (3) Diacetyl isopropylamine
 - (4) Monoacetyl isopropylamine
- 72. Which one of the following is used to make nonstick cookware?
 - (1) Polystyrene
 - (2) Polyvinyl chloride
 - (3) Poly ethylene terephthalate
 - (4) Poly tetrafluoroethylene

- 73. Isobutyl bromide reacts with aq. KOH to give (as major product)
 - (1) n-Butyl alcohol (2) t-Butyl alcohol
 - (3) Isobutyl alcohol (4) Isobutene
- 74. Phenyl salicylate can be used as
 - (1) Antipyretic (2) Antiseptic
 - (3) Disinfectant (4) Analgesic
- 75. Pyrolusite is an ore of
 - (1) Mg (2) Mn
 - (3) Fe (4) Zn
- 76. Physical adsorption is favourable at
 - (1) High temperature
 - (2) Low temperature
 - (3) Room temperature
 - (4) Independent of temperature
- 77. Be and Al differ in
 - (1) Exhibiting maximum covalency in compounds
 - (2) Showing amphoteric nature in their oxides
 - (3) Forming covalent halides
 - (4) Forming polymeric halides
- 78. The number of σ -bonds in P_4O_{10} is
 - (1) 12 (2) 14
 - (3) 16 (4) 20
- 79. Which one of the following has a regular tetrahedral structure?
 - (1) $[Ni(CN)_{4}]^{2-}$ (2) BF₄
 - (3) SF₄ (4) XeF₄
- 80. The products formed by the disproportionation reaction of hypochlorous acid are
 - (1) $HCIO_2$ and $HCIO_4$ (2) HCl and HClO₂
 - (3) HCl and Cl₂O (4) HClO₃ and Cl₂O
- 81. CO₂ and NH₃ react at 473 K and 220 atmospheric pressure to form
 - (1) N_2O and CO (2) $HCONH_2+H_2O$
 - (3) CH_3CONH_2 (4) Urea and H₂O
- 82. Which of the following is an acidic hydride of nitrogen?
 - (1) HN_{3} (2) NH₃
 - (4) N_2H_2 (3) N_2H_4
- 83. Ozone oxidises iodine in the presence of moisture to
 - (1) Hydroiodic acid (2) Periodic acid
 - (3) Iodic acid (4) Hypoiodous acid

- 84. Which pair of metals becomes passive in conc. HNO₃?
 - (1) Fe, Al (2) Cu, Pt
 - (3) Ni, Mg (4) Sn, Ni
- 85. CuCl is sparingly soluble in water but it dissolves in KCl solution due to the formation of
 - (1) $K_3[CuCl_4]$ (2) $K_2[CuCl_4]$
 - (3) K₃[CuCl₂] (4) $K_2[CuCl_2]$
- 86. Silver chloride dissolves in NH₄OH forming a compound 'X'. Which of the following is true about 'X'?
 - (1) X is a complex compound
 - (2) X is a diamagnetic in a nature
 - (3) X is a cationic complex
 - (4) All of these
- 87. Hexafluorocobaltate (III) ion is a high spin complex where the hybrid state of cobalt is
 - (1) $sp^{3}d$ (2) $sp^{3}d^{2}$
 - (3) $d^2 s p^3$ (4) dsp^2
- 88. When an organic compound containing phosphorous is fused with fusion mixture, it forms
 - (1) Na_2HPO_4 (2) Na_3PO_4
 - (3) NaH₂PO₄ (4) Na₃PO₃
- 89. Product OH+ N₂⁺Cl⁻· The product can be



- (4) Both (1) & (2)
- 90. Which of the following compound will be most reactive towards electrophilic substitution reaction?



BOTANY

- 91. Complete oxidation of organic substances in the presence of oxygen
 - (1) Is most common in higher organisms
 - (2) Produces alcohol, CO₂ and energy
 - (3) Releases CO₂, water and small amount of energy
 - (4) Produces lactic acid and large amount of energy
- 92. When the electrons pass from NADH to oxygen via
 - A in the mitochondrial ETS, they are coupled

to \underline{B} for the production of ATP from ADP and inorganic phosphate.

(1) A – Complex I to V

B - ATP synthase

- (2) A Complex V
 - B Complex I to IV
- (3) A Complex I to IV
 - B ATP synthase
- (4) A Complex I to III
 - B Complex IV and V
- From tips of coleoptiles of oat seedlings, F.W. Went isolated
 - (1) Cytokinin
 - (2) An intercellular factor
 - (3) Gibberellic acid
 - (4) An intracellular factor
- 94. Examine the figure given below and select the right option giving both of the parts (A & B) **correctly** identified



- (1) A Action spectrum
 - B Absorption spectrum of chl-a
- (2) A Absorption spectrum of β -carotene
 - B Action spectrum
- (3) A Action spectrum
 - B Absorption spectrum of β -carotene
- (4) A Absorption spectrum of chl-a
 - B Action spectrum
- 95. Nucleolus disappears and the nuclear envelope breaks down by the end of
 - A Diakinesis B Prophase-II
 - C Telophase D Pachytene
 - (1) Only A (2) A, B
 - (3) Only B (4) C, D
- 96. Evolution of C₄ photosynthetic system is one of the strategies for
 - Maximising water loss as well as availability of CO₂
 - (2) Minimising availability of CO₂ while maximising water loss
 - (3) Minimising water loss as well as availability of CO₂
 - (4) Maximising availability of CO₂ while minimising water loss
- 97. Read the following statement having two blanks (A & B)

"In the case of <u>A</u> the F_1 generation resembles both parents but in <u>B</u> the F_1 resembled either of the two parents."

The one correct option for the two blanks is

- (1) A Dominance
 - B Incomplete dominance
- (2) A Incomplete dominance
 - B Co-dominance
- (3) A Co-dominance
 - B Dominance
- (4) A Co-dominance
 - B Incomplete dominance

- 98. Heterogamety is shown by
 - (1) Male grasshopper (2) Female chicks
 - (3) Female fruit fly (4) Both (1) & (2)
- 99. Which of the following disease shows its transmission from unaffected carrier female to some of the male progeny?
 - (1) Haemophilia
 - (2) Myotonic dystrophy
 - (3) Turner's syndrome
 - (4) Down's syndrome
- 100. How many of the following organisms are chemosynthetic autotrophs?
 - I. Nostoc
 - II. Lactobacillus
 - III. Vibrio
 - IV. Nitrobacter
 - (1) Three (2) Four
 - (3) Two (4) One
- 101. In plants, symptoms like mosaic formation, leaf rolling and vein clearing are produced by
 - (1) Bacteria (2) Nucleoprotein particles
 - (3) Prions (4) Nematodes
- 102. Ascospores, sporangiospores and basidiospores all the three are
 - (1) Produced in sexual reproduction
 - (2) Produced during reproduction in fungi
 - (3) Produced in asexual reproduction
 - (4) Produced after gametic meiosis
- 103. In which of the following part(s) of sunflower, vascular cambium is completely secondary in origin?
 - (1) Root, stem (2) Flower, leaf
 - (3) Root (4) Leaf, stem
- 104. Underground stems of *Colocasia* and ginger are modified to perform
 - (1) Photosynthesis
 - (2) Storage and perenation
 - (3) Respiration and absorption of water
 - (4) Conduction of water and minerals

- 105. Axonemal microtubules are arranged as 9 + 2 array in the cilia or flagella of
 - (1) Agaricus and morels
 - (2) T₂ phage and Vibrio
 - (3) Paramoecium and Euglena
 - (4) Porphyra and Polysiphonia
- 106. Mark the mis-matched pair
 - (1) Medicine Aloe, muliathi
 - (2) Edible oil Soyabean, groundnut
 - (3) Ornamentals Lupin, Petunia
 - (4) Dyes Sesbania, Trifolium
- 107. Plants with great ecological importance but little economic importance belong to
 - (1) Bryophytes (2) Pteridophytes
 - (3) Gymnosperms (4) Algae
- 108. Mark the wrongly matched pair
 - (1) Yeast Budding
 - (2) Date palm Dioecious plant
 - (3) Agave Bulbil
 - (4) *Strobilanthus* External fertilisation
- 109. Flowers are not very colourful and do not produce nectar in
 - A. Entomophily
 - B. Anemophily
 - C. Hydrophily
 - D. Ornithophily
 - (1) B, C (2) A, B
 - (3) C, D (4) A, D
- 110. Which one of the following statements is **true** in respect of post fertilisation structures and events?
 - (1) Perisperm is residual and persistent endosperm in seed
 - (2) Zygote gives rise to the proembryo and subsequently to the heart-shaped, globular and mature embryo in dicots
 - (3) In monocots hypocotyl has a shoot apex and a few leaf primordia enclosed in a hollow foliar structure
 - (4) Transformation of ovules into seed and ovary into fruit proceeds simultaneously

- 111. Find out the hybrid varieties of crop plants which have been developed in India
 - A. TN-1
 - B. HD 1553
 - C. IR-8
 - D. P 1542
 - (1) B, C, D
 - (2) B, D
 - (3) A, B, D
 - (4) A, C

113.

112. Which one of the following is mis-matched pair?

(1)	Biomagnification	 DDT in aquatic food chain 		
(2)	Integrated organic farming	 Cyclical zero waste procedure 		
(3)	Sanitary landfills	 Solid wastes 		
(4)	Waste water	 Use of phenyl 		
There are several ways of removing particulate matter; the most widely used of which is				

- (1) Catalytic converter (2) ESP
- (3) Bag filters (4) Scrubber
- 114. The historic convention on Biological Diversity held in
 - (1) Rio de Janeiro, 2002
 - (2) Brazil, 1992
 - (3) South Africa, 2002
 - (4) Johannesburg, 1992
- 115. Large quantity of protein rich food within a short duration is possible by growing
 - (1) Sequoia (2) Spirulina
 - (3) Toadstool (4) *Triticum*
- 116. The dough, which is used for making bread and idli is fermented by
 - (1) Heterotrophic microbes
 - (2) Bacteria and fungi respectively
 - (3) Autotrophic microbes
 - (4) Fungi and BGA respectively

- 117. Find **correct** one (w.r.t. defining features of all life forms)
 - (1) Cellular organisation
 - (2) Extrinsic growth
 - (3) Reproduction
 - (4) Both (1) & (2)
- 118. Which of the following pair is wrongly matched?
 - (1) Ethephon Fruit ripening in apples
 - (2) Bakane disease of rice seedlings BAP
 - (3) Vernalisation Chilling treatment
 - (4) NAA Synthetic auxin
- 119. Reserve food is very similar to amylopectin and glycogen in structure in
 - (1) Dictyota, Fucus
 - (2) Polysiphonia, Porphyra
 - (3) Volvox, Cladophora
 - (4) Euglena, Ulothrix
- 120. Pinnate leaves persist for a few years in a gymnosperm with
 - (1) Sunken stomata
 - (2) Branched stem
 - (3) Coralloid roots
 - (4) Fungal roots
- 121. deATP, deCTP, deGTP and deTTP serve as
 - (1) Source of energy for polymerisation of amino acids
 - (2) Substrates in replication of DNA
 - (3) Source of energy for transcription
 - (4) Both (1) & (2)
- 122. If the sequence of the coding strand in a transcription unit is written as follows

5'-TGAACTGTAGCATGC-3'

Find out the correct sequence in m-RNA.

- (1) 5'-UGAACUGUAGCUUGC-3'
- (2) 3'-CGUACGAUGACAAGU-5'
- (3) 3'-CGUACGAUGUCAAGU-5'
- (4) 5'-UGACUGUAGCUUGC-3'

- 123. Find out the biomolecule having structural specialities to read the genetic code.
 - (1) Amino acids
 - (2) mRNA
 - (3) Adapter molecule
 - (4) Peptidyl transferase
- 124. RuBisCO as well as PEPcase both are found in the leaves of
 - (1) C₃ plants
 - (2) All photosynthetic plants
 - (3) Dry tropical regions
 - (4) More than one option is correct
- 125. Pseudomonas and Thiobacillus perform the
 - (1) Reduction of nitrate in soil
 - (2) Denitrification and ammonification respectively
 - (3) Oxidation of ammonia in soil
 - (4) Nitrogen fixation and nitrification respectively
- 126. Read the following statements
 - A. Cytoskeleton is involved in mechanical support and motility of the bacterial cell.
 - B. In plants cells, centrioles form spindle apparatus during cell division.
 - (1) Both A & B are incorrect
 - (2) Only A is correct
 - (3) Only B is correct
 - (4) Both A & B are correct
- 127. Mature or ripened ovary of mango and coconut resemble in presence of
 - (1) Double endosperm
 - (2) Scutellum
 - (3) Stony hard endocarp
 - (4) Undifferentiated pericarp
- 128. Permanent tissues having all cells similar in structure and function is
 - (1) Collenchyma, sclerenchyma
 - (2) Xylem, phloem
 - (3) Parenchyma, xylem
 - (4) Phloem, collenchyma

- 129. Which one of the following is an **incorrect** statement?
 - (1) Root of *Pinus* establishes an obligate association with fungi
 - (2) Translocation in phloem is explained by the pressure flow hypothesis
 - (3) Ions are absorbed from the soil mainly by active transport
 - (4) More the solute molecules in the solution, the higher is the osmotic potential
- 130. The stage between meiosis I and meiosis II is
 - (1) Called as interkinesis
 - (2) Generally short lived
 - (3) Followed by Prophase II
 - (4) More than one option
- 131. Environmental plasticity is found in
 - (1) Cotton (2) Buttercup
 - (3) Larkspur (4) Coriander
- 132. Anther wall layer meant for synthesis of sporopollenin and incompatibility proteins is
 - (1) Endothecium
 - (2) Middle layers
 - (3) Tapetum
 - (4) Tapetum and middle layers
- 133. Among plants, maximum species global diversity belong to a taxon, known as
 - (1) Algae (2) Angiosperms
 - (3) Mosses (4) Lichens
- 134. Pyramid of energy in a forest ecosystem considering GFC is
 - (1) Spindle shaped (2) Inverted
 - (3) Upright (4) Urn shaped
- 135. Read the following statements
 - A. A considerable amount of NPP is utilised by plants in respiration.
 - B. Trees occupy top vertical strata of a forest, herbs and grasses the second and shrubs occupy the bottom layers.
 - (1) Both A & B are correct
 - (2) Only A is correct
 - (3) Only B is correct
 - (4) Both A & B are incorrect

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- 136. How many of the given statements are wrong?
 - A. The rate of diffusion of gas at the respiratory membrane depends upon its solubility as well as on the thickness of membrane.
 - B. The amount of CO_2 that can diffuse through the diffusion membrane per unit difference in partial pressure is much higher when compared to that of O_2 .
 - C. Oxygen can bind with haemoglobin in an irreversible manner to form oxyhaemoglobin.
 - D. The factors responsible for formation of oxyhaemoglobin include high pO₂, low pCO₂, high H⁺ concentration and low temperature.
 - (1) One (2) Two
 - (3) Three (4) Four
- 137. Consider the following statements A–D with certain blanks and choose the option which **correctly** fills up these blanks.
 - A. A special neural centre in the (i) can moderate the cardiac function through autonomic nervous system.
 - B. (ii) increase cardiac output.
 - C. The ventricular systole starts shortly after (iii)
 - D. The condition of erythroblastosis foetalis can be avoided by administering <u>(iv)</u> to the mother immediately after delivery of first child.
 - (1) (i) Cerebrum
 - (ii) Sympathetic nerves
 - (iii) Q wave
 - (iv) Anti -Rh antibodies
 - (2) (i) Medulla Oblongata
 - (ii) Parasympathetic nerves
 - (iii) R wave
 - (iv) Rh antibodies
 - (3) (i) Pons
 - (ii) Sympathetic nerves
 - (iii) P wave
 - (iv) Rh antibodies
 - (4) (i) Medulla Oblongata
 - (ii) Adrenal medullary hormones
 - (iii) Q wave
 - (iv) Anti -Rh antibodies

- 138. Find out the **incorrect** statement w.r.t. generation conduction of nerve impulse.
 - The resting axonal membrane is comparatively more permeable to potassium ions and nearly impermeable to sodium ions
 - (2) In the resting membrane the outer surface of the axonal membrane possesses a positive charge while its inner surface is negatively charged and therefore, is polarised
 - (3) The membrane becomes depolarised due to rapid influx of Na⁺ on applying threshold stimulus
 - (4) Depolarisation is followed by repolarisation in which rapid efflux of Na⁺ takes place restoring the resting potential of the membrane at the site of excitation
- 139. Following is a diagrammatic representation of endocrine glands with a labelled part 'A'. The part 'A' serves as the connection between



- (1) Hypothalamus and adenohypophysis
- (2) Hypothalamus and neurohypophysis
- (3) Hypothalamus and pars nervosa
- (4) Hypothalamus and epiphysis
- 140. All of the following are functions of Catecholamines, **except**
 - (1) Pupilary dilation
 - (2) Piloerection
 - (3) Tachycardia
 - (4) Lipogenesis

- 141. Given below are four statements A–D w.r.t. mechanism of muscle contraction. Choose from the options stating them as true (T) and false (F)
 - A. The action potential in the sarcolemma is generated by release of calcium ions in the sarcoplasm.
 - B. The globular head of meromyosin possesses ATPase enzyme activity which causes hydrolysis of ATP and the energy derived make myosin head bind to exposed active sites on actin.
 - C. Cross bridge formation is followed by pulling the attached myosin filaments towards the centre of 'I' band.
 - D. The breaking of cross bridge requires binding of new ATP at the myosin head.

	Α	В	С	D
(1)	Т	Т	F	Т
(2)	F	т	F	F
(3)	F	Т	F	Т
(4)	Т	F	F	Т

142. Find out the **correct** labelling of bones A, B, C & D in the given figure



B - Humerus

D - Radius

D - Ulna

D - Ulna

- (1) A Clavicle
 - C Ulna
- (2) A Clavicle B Humerus
 - C Radius
- (3) A Scapula B Humerus
 - C Ulna D Radius
- (4) A Clavicle B Femur
 - C Radius

143. Following is a diagrammatic representation of a nephron showing blood vessels, duct and tubule. These structures have been labelled as A, B. Answer the questions that follow.



- (i) Which parts are located in cortical region of kidney
- (ii) Which structure is highly reduced in cortical nephrons?
- (iii) Identify the part F.
- (iv) Which parts collectively form malpighian body?
- (1) (i) A, C, D, E
 - (ii) B
 - (iii) Collecting duct
 - (iv) A & C
- (2) (i) A, C, D, E
 - (ii) B
 - (iii) Vasa recta
 - (iv) A & C
- (3) (i) B, F, G, H
 - (ii) F
 - (iii) Vasa recta
 - (iv) A & D
- (4) (i) B, F, G, H
 - (ii) F
 - (iii) Collecting duct
 - (iv) C & D
- 144. Which of the following statement is incorrect?
 - Juxta glomerular apparatus is a special sensitive region formed by cellular modifications in the distal convoluted tubule and the afferent arteriole at the location of their contact
 - (2) Nearly 99 percent of the filtrate gets reabsorbed by the renal tubules
 - (3) Nitrogenous wastes in the filtrate get reabsorbed through active transport
 - (4) Human kidneys can produce urine nearly four times concentrated than the initial filtrate formed

(14)

- 145. All of the following digestive enzymes are secreted as pro-enzymes, **except**
 - (1) Pepsin (2) Trypsin

(3) Carboxypeptidase (4) Sucrase

146. Which of the following statements are correct?

- A. The chemical process of digestion is initiated in oral cavity by the hydrolytic action of the carbohydrate splitting enzyme, the salivary amylase.
- B. Lipases are absent in secretion of gastric glands.
- C. Bile helps in emulsification of fats and also activates lipases.
- D. Nucleases in the succus entericus acts on nucleic acids to form nucleotides and nucleosides.
- (1) A, B, C (2) A, C
- (3) C, D (4) A, C, D
- 147. Given figure represents a type of connective tissue with certain labelled structures A, B, C & D



Which of the following option correctly defines the labelled structure?

- (1) A Mast cell Secretes histamine, heparin and serotonin
- (2) B Fibroblast Secrete major amount of matrix
- (3) C Elastic fibres Provides tissue tensile strength
- (4) D Macrophage Engulf germ particle
- 148. Given below is a diagram of muscular tissue. Find the **correct** description of this tissue w.r.t. it's type and location



(1)	Striated and voluntary -	Found closely attached to skeletal bones
(2)	Non-striated and involuntary -	Present only in heart
(3)	Non-striated and involuntary -	Present in wall of blood vessels, stomach and intestine

- (4) Non-striated and voluntary Present in urinary bladder and uterus
- 149. Which of the phylum is a **correct** match w.r.t. phylum and its three examples?
 - (1) Mollusca Pinctada, Loligo, Cucumaria
 - (2) Annelida *Pheretima*, *Hirudinaria*, *Ctenoplana*
 - (3) Arthropoda Laccifer, Limulus, Chaetopleura
 - (4) Cnidaria Pennatula, Meandrina, Aurelia
- 150. Given below are diagrams of four animals A, B, C & D. Find the **correct** option w.r.t. common features of two of these animals.



151. Which of the following animal is **correctly** matched with it's genus and two of its characteristics?







- 153. Which of the following statement is **incorrect** w.r.t. alimentary canal of cockroach?
 - Oesophagus leads into a sac like structure called crop used for storing of food
 - (2) Gizzard has an outer layer of thick circular muscles and thick inner cuticle forming six highly chitinous plate called teeth
 - (3) A ring of 6-8 blind tubules called hepatic or gastric caecae is present at the junction of foregut and midgut
 - (4) The hindgut is narrower than midgut and is differentiated into ileum, colon and rectum
- 154. Consider the following features
 - A. Exclusively marine, radially symmetrical, diploblastic organisms with tissue level of organisation.
 - B. The body bears eight external rows of ciliary comb plates.
 - C. Digestion is both extra cellular and intra cellular.
 - D. Exhibit Bioluminescence.
 - E. Monoecious, fertilisation external with indirect development.

The above mentioned features belong to the animal

- (1) Physalia
- (2) Euspongia
- (3) Fasciola
- (4) Pleurobrachia

155. Match the following structures of cockroach in column I with their location provided in column II.

	Column I	Column II			
a.	Testes in males	(i)	6 th abdominal segment		
b.	Mushroom gland	(ii)	2 nd -6 th abdominal segment		
C.	Spermatheca	(iii)	6 th -7 th abdominal segment		
d.	Ovaries in females	(i∨)	4 th -6 th abdominal segment		
(1)	a(ii), b(iii), c(i), d(iv)	(2)	a(iv), b(iii), c(i), d(ii)		
(3)	a(ii), b(i), c(iii), d(iv)	(4)	a(iii), b(i), c(iv), d(ii)		
Wh	ich of the following is	an	incorrect match w		

156. Which of the following is an **incorrect** match w.r.t. organic compound shown and its identification?



- A nitrogenous base, adenine
- 157. How many of the given statements are correct?
 - A. Water is the most abundant chemical in the living organisms.
 - B. A protein is a heteropolymer and not a homopolymer.
 - C. In a polysaccharide, left end is called reducing end and the right end is called non-reducing end.
 - D. A nucleotide is the building block of nucleic acids, which consist of a heterocyclic compound (nitrogenous base), a monosaccharide and a phosphate.
 - (1) One (2) Two
 - (3) Three (4) Four

- 158. Which of the following factors can affect the enzymatic activity?
 - A. Change in temperature
 - B. Change in pH
 - C. Change in substrate concentration
 - D. Binding of specific chemical to enzyme
 - (1) A, B, C, D
 - (2) A, B, D
 - (3) A, B
 - (4) B, C, D
- 159. Consider the following statements A-D with certain blanks. Find the option which correctly fills up these blanks
 - A. The protein portion of enzyme is called (i) .
 - B. Prosthetic groups are (ii) bound to the protein part of enzyme.
 - (iii) is the prosthetic group of enzyme C. catalase.
 - (iv) are nucleic acids with catalytic power. D.
 - (ii) Tightly (1) (i) Co-factor
 - (iii) Haem (iv) Ribozyme
 - (2) (i) Apoenzyme (ii) Loosely
 - (iv) Ribonuclease P (iii) NAD
 - (ii) Loosely (3) (i) Co-factor
 - (iii) NAD

(iii) Haem

- (iv) Ribonuclease P
- (ii) Tightly (4) (i) Apoenzyme
 - (iv) Ribozyme
- 160. The given reproductive structure is a method of



- (1) Asexual reproduction in Hydra
- (2) Sexual reproduction in earthworm
- (3) Asexual reproduction in sponges
- (4) Sexual reproduction in cockroach

161. Following is diagrammatic representation of menstrual cycle depicting certain events. The uterine events have been labelled as phases A, B, C & D. The ovarian hormones have been labelled as W, X and pituitary hormones as Y and Z. Answer the questions that follow.



- (i) Lack of which phase is indicative of pregnancy?
- (ii) During which phase implantation of blastocyst takes place?
- (iii) Identify the hormones W, X, Y & Z.
- (iv) A preovulatory surge of which hormone is necessary to cause ovulation?
- (1) (i) Phase A (2) (i) Phase A
 - (ii) Phase B(ii) Phase D(iii) W Estrogen(iii) W Estrogen
 - X Progesterone X Progesterone Y - FSH Y - FSH Z - LH Z - LH
 - (iv) Z (iv) Z
- (3) (i) Phase D (4) (i) Phase D
 - (ii) Phase C (ii) Phase B
 - (iii) W Estrogen
 X Progesterone
 Y LH
 Z FSH
 Z LH

(iv) Y

(iv) Y

- 162. Find the **incorrect** statement w.r.t. pregnancy and embryonic development in humans.
 - Placenta is a functional unit between developing embryo and maternal body formed by interdigitation of chorionic villi and uterine tissue
 - (2) Human chorionic gonadotropin(hCG), human placental lactogen(hPL) and relaxin are produced in women only during pregnancy
 - (3) The inner cell mass contain stem cells which have the potency to give rise to all the tissues and organs
 - (4) By the end of second trimester, the first movements of the foetus and appearance of hair on head is usually observed
- 163. According to 2001 census report, the population growth rate of India was
 - (1) 17/1000/year
 - (2) 25/1000/year
 - (3) 33/1000/year
 - (4) 8/1000/year

(2) LNG-20

(3) CuT

(4) Implant

164. Which of the following is **correct** description of the contraceptive device shown?



- (1) Lippes loop Increase phagocytosis of sperms in uterus
 - Make uterus unsuitable for implantation and cervix hostile to sperms
 - Cu ions released suppress sperm motility and the fertilising capacity of sperms

 Inhibit ovulation and implantation as well as alter the quality of cervical mucus to retard entry of sperms

- 165. How many of the following statements are wrong w.r.t. STDs?
 - A. AIDs is the most dangerous STD.
 - B. Hepatitis-B, genital herpes and HIV infection are not curable
 - C. Ectopic pregnancy and cancer are not included under the complications of STDs.
 - D. The incidence of STDs are reported to be very high among persons in the age group of 15-24 vears.
 - (2) Two (1) One
 - (3) Three (4) Four
- 166. Which of the following ART involves transfer of an ovum collected from a donor into the fallopian tube of another female who cannot produce one, but can provide suitable environment for fertilisation and further development?

- (3) ICSI (4) GIFT
- 167. Which of the following is correct match w.r.t. human ancestor, its cranial capacity and certain feature?

	Ancestor	Cranial capacity	Feature
(1)	Australopithecus	500CC	Hunted with stone weapons but essentially ate fruit
(2)	Homo habilis	650-800CC	First human like being the hominid. They probably ate meat
(3)	Homo erectus	900CC	Probably did not ate meat

- (4) More than one option is correct
- 168. Which of the following factor can change allelic frequency?
 - A. Gene flow
 - B. Natural selection
 - C. Random mating
 - D. Founder effect
 - E. Genetic recombination
 - (1) A, B, C, D, E (2) A, B, E
 - (3) C, D, E (4) A, B, D, E

169. The following diagram shows forelimbs of certain mammals. These represent



- - Divergent evolution

Convergent evolution

- Divergent evolution
- (3) Homologous organs (4) Analogous organs

(2) Analogous organs

- 170. Based on observations made during a sea Voyage in a sail ship called H.M.S. Beagle, Charles Darwin concluded that
 - (1) Evolution of life forms had occurred driven by use and disuse of organs
 - (2) Allelic frequencies in a population are stable and is constant from generation to generation
 - (3) Mutation causes evolution as these are random and directionless
 - (4) Existing life forms share similarities to varying degrees not only among themselves but also with life forms that existed millions of years ago
- 171. Which of the following is a correct match w.r.t. pathogen and its site of attack (body cell/organ) in the human body?
 - (1) Ascaris Lymphatic vessels
 - (2) Trichophyton -Lungs
 - (3) Plasmodium -Hepatocytes and RBCs
 - Small intestine (4) Entamoeba histolytica
- 172. All of the following statements are correct w.r.t. heroin, except
 - (1) It binds to specific opioid receptors present in our central nervous system and gastrointestinal tract
 - (2) It is chemically diacetylmorphine
 - (3) It is formed by ethylation of cannabinoids
 - (4) It is a depressant and slows down body functions

(19)

173. Fill in the blank

Α

<u>A</u> involves mating of animals of two different breeds. *Hisardale*, a new breed of <u>B</u> has been developed through this mechanism.

В

- (1) Cross-breeding Sheep
- (2) Cross-breeding Goat
- (3) Out-crossing Sheep
- (4) Out-crossing Goat
- 174. Hydrochloric acid, saliva and tears are included in
 - (1) Physical barriers of innate immunity
 - (2) Physiological barriers of innate immunity
 - (3) Cytokine barriers of innate immunity
 - (4) Cellular barriers of innate immunity

175. Find the **incorrect** match

- Colostrum IgA antibodies
 Recombinant DNA Hepatitis B vaccine
- technology produced from yeast
- (3) Allergy Release of chemicals like histamine and serotonin from the mast cells
- (4) Auto-immune Adrenaline disease
- 176. Which of the following limitation of traditional hybridisation procedure has been overcome by rDNA technology?
 - Introduction of desirable genes without introducing undesirable genes into the target organism
 - (2) To permit variation in the target organism
 - (3) To produce hybrid vigour
 - (4) Both (2) & (3)

177. Which of the following technique has been shown in the given diagram?



- Agarose gel electrophoresis in which smallest DNA fragment moves farther
- (2) Identification and elimination of non-transformants
- (3) Agarose gel electrophoresis in which DNA fragments resolve according to their size
- (4) More than one option is correct
- 178. In the given *E.coli* cloning vector pBR322, restriction site of *Pvu* II is present in



- 179. All of the following are significance of using GM plants, **except**
 - (1) GM plants are tolerant to abiotic stresses
 - (2) GM plants are less dependent on chemical pesticides
 - (3) GM plants have increased efficiency of mineral usage
 - (4) GM plants help in increasing post harvest losses
- 180. Which of the following recombinant protein has been used to treat emphysema?
 - (1) β -galactosidase (2) Alpha-lactalbumin
 - (3) α -1-antitrypsin
- (4) Adenosine deaminase

ANSWERS

1.	(3)	37.	(3)	73.	(2)	109.	(1)	145.	(4)
2.	(1)	38.	(4)	74.	(2)	110.	(4)	146	(2)
3.	(1)	39.	(1)	75.	(2)	111.	(2)	147.	(2)
4.	(4)	40.	(4)	76.	(2)	112.	(4)	148.	(3)
5.	(1)	41.	(3)	77.	(1)	113.	(2)	149.	(4)
6.	(3)	42.	(4)	78.	(3)	114.	(2)	150.	(3)
7.	(3)	43.	(2)	79.	(2)	115.	(2)	151.	(3)
8.	(2)	44.	(3)	80.	(2)	116.	(1)	152.	(1)
9.	(4)	45.	(2)	81.	(4)	117.	(1)	153.	(4)
10.	(2)	46.	(3)	82.	(1)	118.	(2)	154.	(4)
11.	(1)	47.	(3)	83.	(3)	119.	(2)	155.	(2)
12.	(3)	48.	(4)	84.	(1)	120.	(3)	156.	(3)
13.	(1)	49.	(4)	85.	(1)	121.	(2)	157.	(3)
14.	(3)	50.	(2)	86.	(4)	122.	(3)	158.	(1)
15.	(2)	51.	(3)	87.	(2)	123.	(3)	159.	(4)
16.	(4)	52.	(3)	88.	(2)	124.	(3)	160.	(3)
17.	(1)	53.	(2)	89.	(4)	125.	(1)	161.	(2)
18.	(1)	54.	(2)	90.	(2)	126.	(1)	162.	(4)
19.	(1)	55.	(1)	91.	(1)	127.	(3)	163.	(1)
20.	(4)	56.	(3)	92.	(3)	128.	(1)	164.	(3)
21.	(1)	57.	(3)	93.	(2)	129.	(4)	165.	(1)
22.	(1)	58.	(3)	94.	(4)	130.	(4)	166.	(4)
23.	(2)	59.	(2)	95.	(2)	131.	(2)	167.	(1)
24.	(2)	60.	(4)	96.	(4)	132.	(3)	168.	(4)
25.	(1)	61.	(2)	97.	(3)	133.	(2)	169.	(3)
26.	(1)	62.	(2)	98.	(4)	134.	(3)	170.	(4)
27.	(3)	63.	(2)	99.	(1)	135.	(4)	171.	(3)
28.	(1)	64.	(1)	100.	(4)	136.	(2)	172.	(3)
29.	(3)	65.	(2)	101.	(2)	137.	(4)	173.	(1)
30.	(1)	66.	(4)	102.	(2)	138.	(4)	174.	(2)
31.	(2)	67.	(4)	103.	(3)	139.	(1)	175.	(4)
32.	(2)	68.	(2)	104.	(2)	140.	(4)	176.	(1)
33.	(1)	69.	(2)	105.	(3)	141.	(3)	177.	(4)
34.	(1)	70.	(3)	106.	(4)	142.	(2)	178.	(3)
35.	(1)	71.	(3)	107.	(1)	143.	(1)	179.	(4)
36.	(3)	72.	(4)	108.	(4)	144.	(3)	180.	(3)

(21)